
#### Abstract

New chiral pyrazoles, (4R,7R)-4-methyl-7-isopropyl-3-phenyl- (3-phenylisomenthopyrazole; cis-1), (4R,7S)-4-methyl-7-isopropyl- (l-menthopyrazole; trans-2), (4R,7R)-4-isopropyl-7-methyl- (isocarvomenthopyrazole, cis-3) and (4R,7S)-4-isopropyl-7-methyl-4,5,6,7-tetrahydro- $1 H$-indazole (carvomenthopyrazole, trans-3) were prepared. The diastereomeric pairs of these $\mathbf{1 - 3}$ were structurally characterized by NMR spectroscopy. The subtle differences of structures of $\mathbf{1 - 3}$ should induce the useful effects for a chiral auxiliary or a chiral catalyst.


J. Heterocyclic Chem., 39, 1235(2002).

We have previously developed the preparation and the utilities of ( $4 R, 7 S$ )-3-phenyl-4-methyl-7-isopropyl-4,5,6,7-tetrahydro- 1 H -indazole (3-phenyl- $l$-menthopyrazole; trans-1) as a chiral auxiliary [1]. This auxiliary is easily prepared from commercially available $l$-menthone in good yield with high optical purity, and has unique structure and properties relative to the conventional chiral auxiliaries [2]. The most important characteristics of this auxiliary are that the substrate terminates to nitrogen atom of heteroaromatic pyrazole ring and that the 3-phenyl ring of trans $\mathbf{- 1}$ is close to the ( $4 R$ )-methyl group. Subsequently, the steric repulsion is relaxed by rotation of the phenyl ring, and the substrate is located in the chiral environment. This structural feature causes the diastereofacial attack on the substrate moiety in the reactions with alkyl halides [3], diphenyldisulfide [4], acyl chloride [5], aldehydes [6], and $\mathrm{C}=\mathrm{N}$ compounds [7]. Moreover, the asymmetric additions of Grignard reagents [8], dienes [9] and 1,3-dipolar compounds [10] on 2-( $\alpha, \beta$-unsaturated)acyl-3-phenyl- $l$-menthopyrazoles have been reported. Otherwise, the extensive use of these optically active pyrazoles was studied as the chiral catalyst for the asymmetric borane reduction of ketones $[11,12]$ and the asymmetric dialkylzinc addition on aldehydes [13].

| $\mathrm{R}^{2}$ | 3-Phenyl-l-menthopyrazole (trans-1): | $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=(R)-\mathrm{Me}, \mathrm{R}^{3}=(S)-\mathrm{Pr}$ |
| :---: | :---: | :---: |
|  | 3-Phenylisomenthopyrazole (cis-1): | $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=(R)-\mathrm{Me}, \mathrm{R}^{3}=(R)-i \mathrm{Pr}$ |
|  | $l$-Menthopyrazole (trans-2): | $=\mathrm{H}, \mathrm{R}^{2}=(R)-\mathrm{Me}, \mathrm{R}^{3}=(S)-$ |
|  | Isomenthopyrazole (cis-2): | $\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=(R)-\mathrm{Me}, \mathrm{R}^{3}=(R)-i \mathrm{Pr}$ |
| $\mathrm{R}^{3} \quad \mathrm{H}$ | C | ${ }^{1}=\mathrm{H}, \mathrm{R}^{2}=(R)-i \mathrm{Pr}, \mathrm{R}^{3}=(S)-\mathrm{Me}$ |
|  | Isocarvomenthopyrazole (cis-3): | $\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=(\mathrm{R})-\mathrm{Pr}, \mathrm{R}^{3}=(R)-\mathrm{Me}$ |

Figure

In the meanwhile, the preparation of $(4 R, 7 R)$-4-methyl-7-isopropyl-4,5,6,7-tetrahydro-1H-indazole (isomenthopyrazole; cis-2) from $l$-menthone was reported, where (4R,7S)-4-methyl-7-isopropyl-4,5,6,7-tetrahydro- 1 H -indazole ( $l$-menthopyrazole; trans-2) was formed in a yield too low to be isolated [14]. Some of cis-2 derivatives were
used as chiral catalysts in the asymmetric reactions such as bis(isomenthopyrazolyl)methanes in allylic alkylation [15], bis(isomenthopyrazolyl)pyridine in cyclopropanation [16] and tris(isomenthopyrazolyl)hydroborate in polymerization of lactides [17].
Because of difficulty in isolating the trans-2, this compound up until now had not been studied or characterized. Also (4R,7R)-4-methyl-7-isopropyl-3-phenyl-4,5,6,7-tetrahydro- $\mathrm{H} H$-indazole (3-phenylisomenthopyrazole; cis1) should perhaps have a different twisting angle of the 3-phenyl ring against the pyrazole ring, relative to that of epimeric trans-1. Therefore, we have desired to investigate the isolation and the characterization of both cis- $\mathbf{1}$ and trans-2, which were expected to exhibit different steric effects as a chiral auxiliary and/or a chiral catalyst. Moreover, (4R,7S)-4-isopropyl-7-methyl-4,5,6,7-tetrahy-dro- 1 H -indazole (carvomenthopyrazole; trans-3) and ( $4 R, 7 R$ )-4-isopropyl-7-methyl-4,5,6,7-tetrahydro- 1 H indazole (isocarvomenthopyrazole; cis-3) attract our interest in their preparation and the characterization. These regioisomers of cis-/trans-2 have a bulky isopropyl group on 3-position but not on 7-position.

In this paper we report the syntheses of new chiral pyrazoles cis-1, trans-2, trans-3 and cis-3, which should exhibit useful properties which make them prospective chiral auxiliary or chiral catalyst in the asymmetric organic synthesis. Also the structural features of diastereomers of 1-3 were revealed by the comparison of their 1D and 2D NMR spectra.

Results and Discussion.
3-Phenylisomenthopyrazole (cis-1) was synthesized from ( $2 R, 5 R$ )-2-isopropyl-5-methylcyclohexanone (isomenthone) according to the preparative method of trans-1 (Scheme 1). Isomenthone was prepared by chromic acid oxidation of commercially available isomenthol. Isomenthone was acylated by benzoyl chloride in the presence of LDA to afford 2-benzoylisomenthone (4), which was treated with hydrazine monohydrate to give cis-1 in moderate yield. During the benzoylation and pyrazole ring
formation, no epimerization and racemization was observed in GC through a chiral stationary-phase column.

13:1. On the contrary, $l$-menthone was formylated with LDA (Scheme 2, Method B) to give the mixture of cis-5

Scheme 1


The syntheses of 3-phenylisomenthopyrazole (cis-1).

As shown in the synthesis of isomenthopyrazole (cis-2) and $l$-menthopyrazole (trans-2), the mixture of $(2 R, 5 R)$ -(cis-5) and ( $2 S, 5 R$ )-6-hydroxymethylene-2-isopropyl-5methylcyclohexanone (trans-5) was prepared by the formylation of ( $2 S, 5 R$ )-2-isopropyl-5-methylcyclohexanone ( $l$-menthone) following a procedure that uses sodium hydride, which was modified from that method by using sodium methoxide [14b] (Scheme 2, Method A). In this formylation, the $l$-menthone skeleton was epimerized into isomenthone skeleton on $C-2$ carbon having $2 R$ configuration, and cis-5 was predominantly formed with the ratio of
and trans-5 with the ratio of 1:3. When isomenthone was formylated according to either Method A or B, cis-5 was predominantly formed with the ratio of $13: 1$, respectively. Namely, the formylation with sodium hydride afforded the cis isomer independent from the configuration of the menthone substrate, while the stereo structure of menthones was partly retained in the formylation using LDA.

Since the separation of diastereomeric mixture of 5 was unsuccessful even by the careful chromatography, the mixture of cis- 5 and trans- 5 was treated with hydrazine hydrate without isolation. Subsequent mixture of

Scheme 2

cis:trans $=13: 1($ Method A)
1:3 (Method B)
The syntheses of menthopyrazoles 2 .

cis:trans $=3: 2($ Method $A)$
1:2 (Method B)
The syntheses of carvomenthopyrazoles 3 .
diastereomeric pyrazoles cis-2 and trans-2 was obtained with the cis/trans ratio depending on that of 5. Although the isolation of cis-2 mixture was formerly accomplished by fractional recrystallization of their hydrochloride salts, the isolation of trans- $\mathbf{2}$ was unsuccessful by either chromatographic separation or fractional recrystallization.
Previously $N$-acylation of trans- $\mathbf{1}$ was studied to give N -acyl-3-phenyl-l-menthopyrazoles, isomers of which were easily purified by chromatographic separation [1]. Moreover, $N$-acyl-3-phenyl- $l$-menthopyrazoles were easily deacylated into trans- $\mathbf{1}$ by the action of nucleophiles such as alcohols [18] and amines [19]. On the basis of these facts, the resolution of the isomeric mixture of cis-2 and trans- $\mathbf{2}$ was undertaken by means of the acylation, chromatographic separation and the subsequent deacylation. When the mixture of cis-2 and trans-2 was treated with acetyl chloride in the presence of triethylamine, acetylation occurred regioselectively at the $\mathrm{N}-2$ position to afford a mixture of cis-6 and trans-6. After chromatographic separation between cis-6 and trans-6, the independent deacetylation of cis-6 and trans-6 gave cis-2 and trans-2 without epimerization by the action of sodium hydroxide or sulfuric acid, respectively. Both of resulting cis-2 and trans- $\mathbf{2}$ were proven to be optically pure by chiral GC.

In order to clarify the substituent effect of methyl and isopropyl groups on $l$-menthopyrazoles and isomenthopyrazoles, the synthesis of (4R)-4-isopropyl-7-methyl-4,5,6,7-tetrahydro- 1 H -indazole (carvomenthopyrazole, 3),
which was the regioisomers of menthopyrazoles, was attempted from ( $R$ )-(-)-carvone ((5R)-2-methyl-5-propenyl-2-cyclohexenone) (Scheme 3). Commercially available ( $R$ )-(-)-carvone was hydrogenated over $\mathrm{Pd} / \mathrm{C}$ to give carvomenthone (7) as 1:2 cis/trans mixture, which were the regioisomers of isomenthone and $l$-menthone. The cis/trans mixture of 7 was formylated by sodium hydride (Method A) or LDA (Method B) to afford a mixture of $(4 S, 7 S)$ - (cis-8) and ( $4 S, 7 R$ )-2-hydroxymethylene-6-methyl-3-isopropylcyclohexanone (trans-8) in the ratios of 3:2 (Method A) and 1:2 (Method B), respectively. Since the isomeric separations of $\mathbf{7 , 8}$ and $\mathbf{3}$ were unsuccessful by chromatography, the mixture of cis-8 and trans-8 was converted into cis- $\mathbf{3}$ and trans- $\mathbf{3}$ by hydrazine monohydrate without any purification. Further the isomeric purification of $\mathbf{3}$ was carried out by way of acetylation, chromatography and then deacetylation, as described above. Chiral GC was used to determine the optical purities of cis-3 and trans-3.

The stereo-chemical structure of these new chiral pyrazoles were studied by NMR spectroscopy using 1D- and 2D-measurement including HMQC and NOESY technique, as listed in Table 1. Although NMR data for trans$\mathbf{1}$ [1] and cis-2 [14b] have already been reported in papers, we took notice on the peaks attributable to the $C-5$ and $C-6$ protons. Both of $C-5$ axial and $C-6$ axial protons of trans1 are triplets of doublets of doublets due to two trans-diaxial anti vicinal couplings ( $J=6.7 \mathrm{~Hz}$ ), an axial-equatorial

Table 1
${ }^{1} \mathrm{H}$ NMR Spectral Data of $\mathbf{1 - 3}$
Chemical shift ( $\delta$ )

|  | cis-1 | trans-1 | cis-2 | trans-2 | cis-3 | trans-3 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | --- | --- | 7.35 (s) | 7.35 (s) | 7.34 (s) | 7.35 (s) |
| 4 | $\begin{gathered} 3.22 \\ (\mathrm{dq}, 12.7,6.4) \end{gathered}$ | $\begin{gathered} 3.09 \\ (\operatorname{sex}, 6.7) \end{gathered}$ | 1.43-1.53 (m) | $\begin{gathered} 2.65 \\ \text { (d-quint, 11.1, 6.0) } \end{gathered}$ | $\begin{gathered} 2.49 \\ (\mathrm{q}, 6.4) \end{gathered}$ | $\begin{gathered} 2.62 \\ (\mathrm{dt}, 10.3,5.1) \end{gathered}$ |
| 5 eq | 1.67-1.73 (m) | $\begin{gathered} 2.04 \\ (\mathrm{dtd}, 13.3,6.7,2.6) \end{gathered}$ | 1.43-1.53 (m) | $\begin{gathered} 1.95 \\ (\mathrm{dm}, 12.9) \end{gathered}$ | 1.57-1.65 (m) | $\begin{gathered} 1.82 \\ (\mathrm{dtd}, 12.4,5.5,2.0) \end{gathered}$ |
| 5 ax | 1.82-1.88 (m) | $\begin{gathered} 1.31 \\ (\mathrm{qd}, 10.6,2.6) \end{gathered}$ | 1.74-1.82 (m) | $\begin{gathered} 1.19 \\ (\mathrm{qd}, 12.9,2.4) \end{gathered}$ | 1.66-1.72 (m) | $\begin{gathered} 1.37 \\ (\mathrm{qd}, 12.7,2.2) \end{gathered}$ |
| 6 eq | 1.67-1.73 (m) | $\begin{gathered} 1.827 \\ (\mathrm{dtd}, 13.2,6.5,2.6) \end{gathered}$ | 1.64-1.69 (m) | $\begin{gathered} 1.88 \\ (\mathrm{dm}, 13.0) \end{gathered}$ | 1.57-1.65 (m) | 1.99-2.07 (m) |
| 6 ax | 1.67-1.73 (m) | $\begin{gathered} 1.56 \\ (\mathrm{qd}, 10.9,2.4) \end{gathered}$ | 1.74-1.82 (m) | $\begin{gathered} 1.46 \\ (\mathrm{qd}, 13.0,2.4) \end{gathered}$ | 1.81-1.85 (m) | $\begin{gathered} 1.31 \\ (\mathrm{qd}, 12.5,2.2) \end{gathered}$ |
| 7 | $\begin{gathered} 2.71 \\ (\mathrm{dt}, 9.1,4.6) \end{gathered}$ | $\begin{gathered} 2.59 \\ (\mathrm{dt}, 7.1,4.5) \end{gathered}$ | $\begin{gathered} 2.61 \\ (\mathrm{q}, 5.8) \end{gathered}$ | $\begin{gathered} 2.74 \\ (\mathrm{dt}, 11.0,5.5) \end{gathered}$ | $\begin{gathered} 2.90 \\ (\operatorname{sex}, 6.4) \end{gathered}$ | $\begin{gathered} 2.78 \\ \text { (d-quin, 10.2, 5.1) } \end{gathered}$ |
| 4-Me | 1.10 (d, 6.9) | 0.99 (d, 6.7) | 0.86 (d, 6.8) | 1.20 (d, 6.8) | --- | --- |
| $7-i \operatorname{Pr}(\mathrm{Me})$ | 0.86 (d, 6.9) | 0.82 (d, 6.8) | 1.04 (d, 6.8) | 0.81 (d, 6.9) | --- | --- |
|  | 1.05 (d, 6.9) | 0.99 (d, 6.6) | 1.18 (d, 6.9) | 1.03 (d, 6.9) | --- | --- |
| $7-i \mathrm{Pr}(\mathrm{CH})$ | 2.26 (oct, 5.8) | 2.16 (oct, 6.6) | 2.11 (oct, 6.5) | 2.27 (broad) | --- | --- |
| $4-\mathrm{iPr}$ (Me) | --- | --- | --- | --- | 0.88 (d, 6.8) | 0.83 (d, 6.9) |
|  | --- | --- | --- | --- | 0.99 (d, 6.8) | 0.99 (d, 6.9) |
| $4-i \mathrm{Pr}(\mathrm{CH})$ | --- | --- | --- | --- | 1.89 (oct, 6.7) | 1.99-2.07 (m) |
| 7-Me | --- | --- | --- | --- | 1.27 (d, 7.0) | 1.301 (d, 6.8) |
| $3-\mathrm{Ph}(\mathrm{p})$ | 7.31 (d, 7.4) | 7.31 (d, 7.4) | --- | --- | --- | --- |
| $3-\mathrm{Ph}(\mathrm{m})$ | 7.39 (d, 7.4) | 7.38 (d, 7.4) | --- | --- | --- | --- |
| $3-\mathrm{Ph}(\mathrm{o})$ | 7.68 (d, 7.5) | 7.60 (d, 7.4) | --- | --- | --- | --- |

geminal coupling ( $J=10.7 \mathrm{~Hz}$ ) and an axial-equatorial vicinal coupling ( $J=2.5 \mathrm{~Hz}$ ). Also those of trans-2 and trans-3 appeared as the peaks of quartet ( $J=12.5 \sim 13.0 \mathrm{~Hz}$ ) of doublet ( $J=2.2 \sim 2.4 \mathrm{~Hz}$ ). These spectral features of trans-1-3 supported their typical half chair form having the two substituent groups on equatorial positions of $C-4$ and $C-7$. On the contrary, both of $C-5$ and $C-6$ protons of cis-2 were observed as the complicated multiplets, where the steric configuration was previously reported to be distorted from the chair form [13-15]. Similarly, the complicated multiplets were attributable to the axial proton signals on $C-5$ and $C-6$ of cis-1 and cis-3. It is suggested that the distorted half-chair six-membered ring of tetrahydroindazole is responsible for the complicated coupling patterns observed for cis-1-3.
The methyl group on $C-4$ position and the $C-4$ proton of trans- $\mathbf{1}$ were shifted to higher and lower fields respectively by the anisotropic effect of the 3-phenyl group, comparing with those of trans-2. This anisotropic observation was supported by the X-ray structural analysis that the benzene ring of trans- 1 was twisted about $40^{\circ}$ against pyrazole ring. In the case of cis-1, the methyl group and the proton signals on $C$-4 were also shifted compared with those of cis-2, but to a lesser extent than those of trans-1. This difference in the magnitude of the anisotropic effect of the 3phenyl group should reflected in the twisting angle of the benzene ring relative to the pyrazole ring of cis- $\mathbf{1}$ compared to that of trans-1, which causes most of the diastereofacial behaviors when used as a chiral auxiliary. In addition, the ${ }^{13} \mathrm{C}$ NMR spectra of cis- $\mathbf{1}$ were found to be the broad signals. This may be explained by a slow tautomerisation between 1 H - and 2 H -tetrahydroindazole. The predominant tautomer seemed to be the 1 H -tetrahydroindazole form in this new chiral pyrazole.
In conclusion, new chiral pyrazoles, $(4 R, 7 R)$-4-methyl-7-isopropyl-3-phenyl- (3-phenylisomenthopyrazole; cis1), (4R,7S)-4-methyl-7-isopropyl- ( $l$-menthopyrazole; trans-2), (4R,7R)-4-isopropyl-7-methyl- (isocarvomenthopyrazole, cis-3) and ( $4 R, 7 S$ )-4-isopropyl-7-methyl-4,5,6,7-tetrahydro- 1 H -indazole (carvomenthopyrazole, trans-3) were prepared. The diastereomeric pairs of these 1-3 were structurally characterized by NMR spectroscopy. The subtle differences of structures of $\mathbf{1 - 3}$ should induce useful effects for use as a chiral auxiliary or chiral catalyst.

## EXPERIMENTAL

Melting points are uncorrected. NMR data were collected on Bruker AVANCE 600 ( 600 MHz ) or JEOL ENM-EX270 (270 MHz ) spectrometer in deuteriochloroform with tetramethylsilane as an internal standard. Optical rotations were observed using a JASCO DIP-370 digital polarimeter. Chiral GC chromatograms were obtained on SHIMADZU GC-14A gas chromatograph using Chrompak Chirasil DEX-CB capillary column ( $\phi 0.25 \mathrm{~mm}$
x 25 m ). THF and $\mathrm{Et}_{2} \mathrm{O}$ were dried over benzophenone ketyl radical, and toluene was distilled over calcium hydride.

Preparation of 3-Phenylisomenthopyrazole.
(3R,6R)-2-Benzoyl-3-methyl-6-isopropylcyclohexanone (4).
To the $\mathrm{Et}_{2} \mathrm{O}$ solution ( 40 ml ) of isomenthol ( $7.8 \mathrm{~g}, 50 \mathrm{mmol}$ ) in ice bath, sodium dichromate dihydrate ( $5.2 \mathrm{~g}, 17 \mathrm{mmol}$ ) in dilute sulfuric acid ( $25 \mathrm{ml}, 7.2 M$ ) was added dropwise over 1 hour. After stirring another hour at room temperature, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic layer was washed with saturated $\mathrm{NaHCO}_{3}$ and saturated NaCl , and dried over anhydrous $\mathrm{MgSO}_{4}$. After removal of the solvent, isomenthone ( 7.2 g , 47 mmol ) was purified by Kugelrohr distillation. The subsequent isomenthone ( $1.5 \mathrm{~g}, 10 \mathrm{mmol}$ ) in THF ( 10 ml ) was added to THF solution ( 10 ml ) of LDA, which was prepared in situ from diisopropylamine ( 1.6 ml ) and butyllithium solution ( $7.5 \mathrm{ml}, 1.59 \mathrm{M}$ in hexane) at $-5^{\circ} \mathrm{C}$. After stirring for 15 minutes at $-5^{\circ} \mathrm{C}$, the mixture was treated with benzoyl chloride ( $1.6 \mathrm{~g}, 11 \mathrm{mmol}$ ) in THF ( 5 ml ), and then warmed up to room temperature with stirring for 1.5 hours. The mixture was quenched with dilute hydrochloric acid and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic layer was washed with saturated NaCl , dried over anhydrous $\mathrm{MgSO}_{4}$, and concentrated. The reaction residue was purified by recrystallization from methanol or hexane; yield $38 \%$; $\mathrm{mp} 108.5-109.5{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 0.87(3 \mathrm{H}, \mathrm{d}$, $\mathrm{J}=6.6 \mathrm{~Hz}), 0.94(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.6 \mathrm{~Hz}), 1.01(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=6.9 \mathrm{~Hz}), 1.54-$ $1.65(1 \mathrm{H}, \mathrm{m}), 1.86-2.02(3 \mathrm{H}, \mathrm{m}), 2.10-2.22(2 \mathrm{H}, \mathrm{m}), 2.63-2.70$ $(1 \mathrm{H}, \mathrm{m}), 4.17(1 \mathrm{H}, \mathrm{d}, \mathrm{J}=7.6 \mathrm{~Hz}), 7.41-7.59(3 \mathrm{H}, \mathrm{m}), 7.88-7.93$ $(2 \mathrm{H}, \mathrm{m})$; ${ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 19.4\left(\mathrm{CH}_{3}\right), 20.4$ $\left(\mathrm{CH}_{3}\right), 21.1\left(\mathrm{CH}_{3}\right), 26.2\left(\mathrm{CH}_{2}\right), 26.9\left(\mathrm{CH}_{2}\right), 28.0(\mathrm{CH}), 36.0$ $(\mathrm{CH}), 56.8(\mathrm{CH}), 64.7(\mathrm{CH}), 128.4(\mathrm{CH}), 128.6(\mathrm{CH}), 133.2$ (CH), 137.5 (C), 197.5 (C), 210.5 (C).

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{O}_{2}$ : C, $79.03 ; \mathrm{H}, 8.58$. Found: C, 79.24; H, 8.55.
(4R,7R)-3-Phenyl-4-methyl-7-isopropyl-4,5,6,7-tetrahydroindazole (cis-1).

The mixture of ( $3 R, 6 R$ )-2-benzoyl-3-methyl-6-isopropylcyclohexanone ( $4,970 \mathrm{mg}, 3.7 \mathrm{mmol}$ ), hydrazine monohydrate ( 2.8 g , 56 mmol ) and hydrazine hydrochloride ( $400 \mathrm{mg}, 5.9 \mathrm{mmol}$ ) in methanol ( 20 ml ) was refluxed for 6 hours. The reaction mixture was quenched with water and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The organic layer was washed with aqueous $\mathrm{NaHCO}_{3}$, and saturated NaCl , dried over anhydrous $\mathrm{MgSO}_{4}$, and concentrated. The reaction residue was purified by recrystallization from aqueous methanol; yield $89 \%$; mp $124-126^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-10.8\left(c 2.73, \mathrm{CHCl}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 17.8\left(\mathrm{CH}_{3}\right), 18.6\left(\mathrm{CH}_{2}\right), 20.4\left(\mathrm{CH}_{3}\right)$, $21.2\left(\mathrm{CH}_{3}\right), 25.8(\mathrm{CH}), 30.0(\mathrm{CH}), 30.3\left(\mathrm{CH}_{2}\right), 39.7(\mathrm{CH}), 119.2$ (C), $126.7(\mathrm{CH}), 127.3(\mathrm{CH}), 128.5(\mathrm{CH}), 133.8(\mathrm{C}), 144.7(\mathrm{C}$, broad), 145.3 ( C , broad). The ${ }^{1} \mathrm{H}$ NMR data were summarized in Table 1 as well as those of trans-1.

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{~N}_{2}$ : C, 80.27; H, 8.72; N, 11.01. Found: C, 79.99; H, 8.72; N, 10.88.

Preparation of cis-2, trans-2, cis-3 and trans-3.
Hydrogenation of ( $R$ )-Carvone.
The $\mathrm{MeOH}(30 \mathrm{ml})$ solution of $(R)$-carvone $(1.05 \mathrm{~g}, 7.0 \mathrm{mmol})$ was hydrogenated over Pd ( $50 \mathrm{mg}, 5 \%$ on carbon) under room pressure and temperature for 4 hours. After filtering off the catalyst, the solvent was removed to give the mixture ( 980 mg ) of cis-

7 and trans-7. The product mixture was purified by Kugelrohr distillation, and was identical with the reported NMR data [20].

General Procedure for Formylation and the Pyrazole Ring Formation.

Method A.
To a suspension of $\mathrm{NaH}\left(1.9 \mathrm{~g}, 48 \mathrm{mmol}, 60 \%\right.$ in oil) in $\mathrm{Et}_{2} \mathrm{O}$ $(40 \mathrm{ml}), l$-menthone or the mixture of cis-7 and trans-7 (42 $\mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(40 \mathrm{ml})$ was added, followed by ethyl formate ( 6.3 $\mathrm{g}, 85 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(40 \mathrm{ml})$. After stirring for 18 hours under argon atmosphere, the mixture was dissolved in $\mathrm{H}_{2} \mathrm{O}$. The organic layer was extracted with $1 M \mathrm{NaOH}$. The combined aqueous layers were acidified with hydrochloric acid, and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The $\mathrm{Et}_{2} \mathrm{O}$ layers were washed with saturated NaCl , and dried over anhydrous $\mathrm{MgSO}_{4}$. After removal of the solvent, the residual oil ( 6.7 g ) was comprised of a $13: 1$ ratio of ( $3 R, 6 R$ )- (cis-5) and ( $3 R, 6 S$ )-2-hydroxymethylene-6-iso-propyl-3-methylcyclohexanones (trans-5). In the case of 7, the residual oil $(2.4 \mathrm{~g})$ was comprised of a 3:2 ratio of $(3 R, 6 R)$ - (cis8) and ( $3 S, 6 R$ )-2-hydroxymethylene-6-isopropyl-3-methylcyclohexanones (trans-8). The mixture was refluxed with hydrazine monohydrate ( $1.1 \mathrm{~g}, 22 \mathrm{mmol}$ ) and hydrazine hydrochloride ( 135 $\mathrm{mg}, 2.0 \mathrm{mmol}$ ) in methanol ( 30 ml ) for 3 hours. The reaction mixture was quenched with $\mathrm{H}_{2} \mathrm{O}$ and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The $\mathrm{Et}_{2} \mathrm{O}$ layer was washed with water and saturated NaCl , dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated. The product ( 6.1 g ) was found to be cis-2 and trans-2 mixture with the ratio of 13:1 after Kugelrohr distillation under reduced pressure. Otherwise, cis-3 and trans- $\mathbf{3}$ mixture ( 2.2 g ) was isolated with the ratio of 3:2 by Kugelrohr distillation under reduced pressure.

## Method B.

To THF solution ( 40 ml ) of LDA which was prepared in situ from diisopropylamine ( 9 ml ) and butyllithium solution ( 40 ml , 1.59 M in hexane), $l$-menthone or the mixture of cis-7 and trans-7 ( 60 mmol ) in THF ( 10 ml ) was added at $-5^{\circ} \mathrm{C}$. After standing for 15 minutes at $-5^{\circ} \mathrm{C}$, ethyl formate ( $6.0 \mathrm{~g}, 81 \mathrm{mmol}$ ) in THF ( 30 $\mathrm{ml})$ was added to the mixture and stirred for 1.5 hours. The mixture was quenched with $\mathrm{H}_{2} \mathrm{O}$ and the organic layer was extracted 1 $M \mathrm{NaOH}$. The combined aqueous layers were acidified with hydrochloric acid, and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The $\mathrm{Et}_{2} \mathrm{O}$ layer was washed with saturated NaCl , and dried over anhydrous $\mathrm{MgSO}_{4}$. After removal of the solvent, the residual oil ( 5.1 g ) was comprised of a 1:3 ratio of cis-5 and trans-5. In the case of 7, the residual oil ( 4.3 g ) was comprised of a $1: 2$ ratio of cis- 7 and trans7. The mixture was refluxed with hydrazine monohydrate ( 1.5 g , 30 mmol ) and hydrazine hydrochloride ( $170 \mathrm{mg}, 2.5 \mathrm{mmol}$ ) in methanol for 3 hours. The reaction mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}$, and the organic layer was washed with water and saturated NaCl , dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated. The product ( 5.6 g ) was found to be cis-2 and trans-2 mixture with the ratio of 1:3 after Kugelrohr distillation under reduced pressure. Otherwise, cis- $\mathbf{3}$ and trans- $\mathbf{3}$ mixture ( 1.6 g ) was isolated with the ratio of $1: 2$ by Kugelrohr distillation under reduced pressure.
Acetylation of cis-2, trans-2, cis-3 and trans-3.
A mixture ( 5 mmol ) of cis-2 and trans-2 or a mixture ( 5 mmol ) of cis-3 and trans-3, prepared as above, was acetylated by acetyl chloride ( 8.8 mmol ) in toluene ( 20 ml ) in the presence of triethylamine ( 15 mmol ), according to the formerly described method
[3]. The subsequent mixture was separated by silica gel column chromatography with benzene-hexane mixture ( $\mathrm{v} / \mathrm{v} 1: 1$ ), respectively. All products were obtained as colorless oils, and cis-6 crystallized upon standing for a few days.
(4R,7S)-2-Acetyl-4-methyl-7-isopropyl-4,5,6,7-tetrahydro-2Hindazole (trans-6).

Trans- 6 showed bp $200{ }^{\circ} \mathrm{C} / 4 \mathrm{mmHg} ;{ }^{1} \mathrm{H}$ NMR ( 270 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 0.88(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.03(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.18$ $(1 \mathrm{H}, \mathrm{q}, J=12.2 \mathrm{~Hz}), 1.21(3 \mathrm{H}, \mathrm{d}, J=6.6 \mathrm{~Hz}), 1.46(1 \mathrm{H}, \mathrm{q}, J=12.2$ $\mathrm{Hz}), 1.90-1.98(2 \mathrm{H}, \mathrm{m}), 2.38-2.45(1 \mathrm{H}, \mathrm{m}), 2.59-2.71(2 \mathrm{H}, \mathrm{m})$, $2.64(3 \mathrm{H}, \mathrm{s}), 7.95(1 \mathrm{H}, \mathrm{d}, J=1.3 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 270 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 18.1\left(\mathrm{CH}_{3}\right), 19.9\left(\mathrm{CH}_{3}\right), 20.9\left(\mathrm{CH}_{3}\right), 21.5\left(\mathrm{CH}_{2}\right), 23.7$ $\left(\mathrm{CH}_{3}\right), 28.1(\mathrm{CH}), 30.1(\mathrm{CH}), 32.2\left(\mathrm{CH}_{2}\right), 41.0(\mathrm{CH}), 123.7$ $(\mathrm{CH}), 128.4(\mathrm{C}), 157.6(\mathrm{C}), 169.5(\mathrm{C})$.

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 70.87 ; \mathrm{H}, 9.15 ; \mathrm{N}, 12.72$. Found: C, 70.48; H, 9.17; N, 12.46.
(4R,7R)-2-Acetyl-4-methyl-7-isopropyl-4,5,6,7-tetrahydro-2Hindazole (cis-6).

Cis-6 showed bp $200{ }^{\circ} \mathrm{C} / 4 \mathrm{mmHg}$; ${ }^{1} \mathrm{H}$ NMR ( 270 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 0.92(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.06(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.19$ $(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.43-1.59(1 \mathrm{H}, \mathrm{m}), 1.65-1.88(3 \mathrm{H}, \mathrm{m}), 2.13-$ $2.28(1 \mathrm{H}, \mathrm{m}), 2.53-2.67(1 \mathrm{H}, \mathrm{m}), 2.64(3 \mathrm{H}, \mathrm{s}), 2.78-2.88(1 \mathrm{H}, \mathrm{m})$, $7.94(1 \mathrm{H}, \mathrm{d}, J=1.3 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 19.0$ $\left(\mathrm{CH}_{3}\right), 20.6\left(\mathrm{CH}_{3}\right), 21.51\left(\mathrm{CH}_{3}\right), 21.58\left(\mathrm{CH}_{2}\right), 22.1\left(\mathrm{CH}_{3}\right), 26.3$ $(\mathrm{CH}), 29.2\left(\mathrm{CH}_{2}\right), 30.6(\mathrm{CH}), 40.0(\mathrm{CH}), 124.0(\mathrm{CH}), 127.6(\mathrm{C})$, 157.5 (C), 169.6 (C).

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 70.87 ; \mathrm{H}, 9.15 ; \mathrm{N}, 12.72$. Found: C, 70.30; H, 9.14; N, 12.70.
(4S,7R)-2-Acetyl-4-isopropyl-7-methyl-4,5,6,7-tetrahydro-2Hindazole (trans-9).

Trans- 9 showed ${ }^{1} \mathrm{H}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 0.84(3 \mathrm{H}, \mathrm{d}$, $J=6.9 \mathrm{~Hz}), 0.99(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.22-1.43(2 \mathrm{H}, \mathrm{m}), 1.34(3 \mathrm{H}$, d, $J=6.9 \mathrm{~Hz}), 1.78-1.94(1 \mathrm{H}, \mathrm{m}), 1.97-2.10(2 \mathrm{H}, \mathrm{m}), 2.56-2.77$ $(2 \mathrm{H}, \mathrm{m}), 2.65(3 \mathrm{H}, \mathrm{s}), 2.78-2.88(1 \mathrm{H}, \mathrm{m}), 7.95(1 \mathrm{H}, \mathrm{d}, J=1.6 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 19.0\left(\mathrm{CH}_{3}\right), 20.6\left(\mathrm{CH}_{3}\right), 21.51$ $\left(\mathrm{CH}_{3}\right), 21.58\left(\mathrm{CH}_{2}\right), 22.1\left(\mathrm{CH}_{3}\right), 26.3(\mathrm{CH}), 29.2\left(\mathrm{CH}_{2}\right), 30.6$ $(\mathrm{CH}), 40.0(\mathrm{CH}), 124.0(\mathrm{CH}), 127.6(\mathrm{C}), 157.5(\mathrm{C}), 169.6(\mathrm{C})$.

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 70.87 ; \mathrm{H}, 9.15 ; \mathrm{N}, 12.72$. Found: C, 70.65; H, 9.15; N, 12.77.
(4S,7S)-2-Acetyl-4-isopropyl-7-methyl-4,5,6,7-tetrahydro-2Hindazole (cis-9).

Cis-9 showed ${ }^{1} \mathrm{H}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 0.89(3 \mathrm{H}, \mathrm{d}$, $J=6.9 \mathrm{~Hz}), 1.00(3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}), 1.29(3 \mathrm{H}, \mathrm{d}, J=6.6 \mathrm{~Hz}), 1.61-$ $1.72(3 \mathrm{H}, \mathrm{m}), 1.76-1.97(2 \mathrm{H}, \mathrm{m}), 2.46-2.54(1 \mathrm{H}, \mathrm{m}), 2.65(3 \mathrm{H}, \mathrm{s})$, 2.85-2.96 ( $1 \mathrm{H}, \mathrm{m}$ ), $7.94(1 \mathrm{H}, \mathrm{d}, J=1.0 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $(270 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 18.6\left(\mathrm{CH}_{3}\right), 20.5\left(\mathrm{CH}_{3}\right), 20.6\left(\mathrm{CH}_{3}\right), 20.9\left(\mathrm{CH}_{2}\right), 21.6$ $\left(\mathrm{CH}_{3}\right), 28.4(\mathrm{CH}), 29.3\left(\mathrm{CH}_{2}\right), 31.2(\mathrm{CH}), 38.5(\mathrm{CH}), 124.1(\mathrm{C})$, 124.4 (CH), 160.1 (C), 169.5 (C).

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}: \mathrm{C}, 70.87 ; \mathrm{H}, 9.15 ; \mathrm{N}, 12.72$. Found: C, 70.69; H, 9.31; N, 12.68.

Deacetylation of N -Acetyl Derivatives.
All 2-acetyl derivatives ( 5.3 mmol ), cis-6, trans-6, cis-9 and trans-9, were refluxed with $\mathrm{NaOH}(1.1 \mathrm{mmol})$ in $\mathrm{MeOH}(15 \mathrm{ml})$, respectively. The resulting mixture was quenched with dil. hydrochloric acid, and extracted with $\mathrm{Et}_{2} \mathrm{O}$. The organic layer was washed with saturated NaCl , dried over anhydrous $\mathrm{MgSO}_{4}$,
and concentrated. The products were purified by the silica gel column chromatography, and cis-2 and trans- $\mathbf{3}$ were crystallized upon standing for a few days. The ${ }^{1} \mathrm{H}$ NMR spectral data were summarized in Table 1.
( $4 R, 7 R$ )-4-Methyl-7-isopropyl-4,5,6,7-tetrahydro- 1 H -indazole (cis-2).

Cis-2 was obtained in $86 \%$ yield; bp $160-165^{\circ} \mathrm{C} / 5 \mathrm{mmHg} ; \mathrm{mp}$ $54-57^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}: 18.45^{\circ}\left(\mathrm{c} 0.084, \mathrm{CHCl}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR ( 270 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 18.5\left(\mathrm{CH}_{3}\right), 20.3\left(\mathrm{CH}_{3}\right), 21.2\left(\mathrm{CH}_{2}\right), 21.8\left(\mathrm{CH}_{3}\right), 25.7$ $(\mathrm{CH}), 29.2\left(\mathrm{CH}_{2}\right), 30.4(\mathrm{CH}), 38.4(\mathrm{CH}), 121.6(\mathrm{C}), 133.1(\mathrm{CH})$, 144.3 (C).
(4R,7S)-4-Methyl-7-isopropyl-4,5,6,7-tetrahydro- 1 H -indazole (trans-2).

Trans-2 was obtained in $95 \%$ yield; bp $180^{\circ} \mathrm{C} / 7 \mathrm{mmHg} ;[\alpha]_{\mathrm{D}}$ -98.1 (c 1.89, $\mathrm{CHCl}_{3}$ ); ${ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 17.7$ $\left(\mathrm{CH}_{3}\right), 20.1\left(\mathrm{CH}_{3}\right), 21.4\left(\mathrm{CH}_{3}\right), 23.3\left(\mathrm{CH}_{2}\right), 27.7(\mathrm{CH}), 30.3$ $(\mathrm{CH}), 32.7\left(\mathrm{CH}_{2}\right), 39.6(\mathrm{CH}), 122.4(\mathrm{C}), 132.1(\mathrm{CH}), 145.2(\mathrm{C})$.
Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2}$ : C, 74.11; H, 10.18; N, 15.71. Found: C, 73.89; H, 10.12; N, 15.76.
(4S,7S)-4-Isopropyl-7-methyl-4,5,6,7-tetrahydro- 1 H -indazole (cis-3).

Cis- $\mathbf{3}$ was obtained in $95 \%$ yield; bp $180^{\circ} \mathrm{C} / 7 \mathrm{mmHg} ;[\alpha]_{\mathrm{D}}$ -9.3 ( c 2.09, $\mathrm{CHCl}_{3}$ ); ${ }^{13} \mathrm{C} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 19.0$ $\left(\mathrm{CH}_{3}\right), 20.8\left(\mathrm{CH}_{3}\right), 20.8\left(\mathrm{CH}_{3}\right), 22.1\left(\mathrm{CH}_{2}\right), 27.3(\mathrm{CH}), 29.5$ $\left(\mathrm{CH}_{2}\right), 31.5(\mathrm{CH}), 38.3(\mathrm{CH}), 118.1(\mathrm{C}), 133.0(\mathrm{CH}), 147.7(\mathrm{C})$.

Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2}$ : C, 74.11; H, 10.18; N, 15.71. Found: C, 73.92; H, 10.18; N, 15.43.
(4S,7R)-4-Isopropyl-7-methyl-4,5,6,7-tetrahydro- 1 H -indazole (trans-3).

Trans- $\mathbf{3}$ was obtained in $95 \%$ yield; $\mathrm{mp} 84-85^{\circ} \mathrm{C} ;[\delta]_{\mathrm{D}}-54.8$ (c $\left.2.07, \mathrm{CHCl}_{3}\right) ;{ }^{13} \mathrm{C}$ NMR ( $270 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 18.0\left(\mathrm{CH}_{3}\right)$, $19.7\left(\mathrm{CH}_{3}\right), 20.2\left(\mathrm{CH}_{3}\right), 24.2\left(\mathrm{CH}_{2}\right), 28.9(\mathrm{CH}), 31.3(\mathrm{CH}), 32.3$ $\left(\mathrm{CH}_{2}\right), 38.9(\mathrm{CH}), 118.4(\mathrm{C}), 132.0(\mathrm{CH}), 148.3(\mathrm{C})$.
Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2}$ : C, 74.11; H, 10.18; $\mathrm{N}, 15.71$. Found: C, 74.16; H, 10.12; N, 15.64.

Acknowledgment.
The authors are grateful to the Chemical Analysis Center, University of Tsukuba, for the measurements of the NMR spectra, and the elemental analyses.

## REFERENCES AND NOTES

[1] C. Kashima, I. Fukuchi, K. Takahashi, and A. Hosomi, Tetrahedron Lett., 34, 8305 (1993).
[2] For recent reviews, see: [a] 'Asymmetric Synthesis', Vol. 1-5, ed. by D. J. Morrison, Academic Press Inc., New York, 19831985; [b] B. H. Kim and D. P. Curran, Tetrahedron, 49, 298 (1993); [c] L. Deloux and M. Srebnik, Chem. Rev., 93, 763 (1993); [d] T. G. Gant and A. I. Meyers, Tetrahedron, 50, 2297 (1994).
[3] C. Kashima, I. Fukuchi, and A. Hosomi, J. Org. Chem., 59, 7821 (1994).
[4] C. Kashima, K. Takahashi, and A. Hosomi, Heterocycles, 42, 241 (1996).
[5] C. Kashima, I. Fukuchi, K. Takahashi, and A. Hosomi, Tetrahedron, 52, 10335 (1996).
[6] C. Kashima, I. Fukuchi, K. Takahashi, K. Fukusaka, and A. Hosomi, Heterocycles, 47, 357 (1998).
[7] C. Kashima, K. Fukusaka, and K. Takahashi, J. Heterocyclic Chem., 34, 1559 (1997).
[8] C. Kashima, K. Takahashi, K. Fukusaka, and A. Hosomi, J. Heterocyclic Chem., 35, 503 (1998).
[9] C. Kashima, K. Fukusaka, K. Takahashi, and Y. Yokoyama, J. Org. Chem., 64, 1108 (1999).
[10] C. Kashima, K. Takahashi, I. Fukuchi, and K. Fukusaka, Heterocycles, 44, 289 (1997).
[11] C. Kashima, Y. Tsukamoto, K. Higashide, and H. Nakazono, J. Heterocyclic Chem., 37, 983 (2000).
[12] C. Kashima, Y. Tsukamoto, Y. Miwa, and K. Higashide, J. Heterocyclic Chem., 38, 601 (2001).
[13] C. Kashima, K. Higashide, Y. Miwa, and Y. Tsukamoto, J. Heterocyclic Chem., in press.
[14] D. D. LeCloux and W. B. Tolman, J. Am. Chem. Soc., 115, 1153 (1993); [b] D. D. LeCloux, C. J. Tokar, M. Osawa, R. P. Houser, M. C. Keyes, and W. B. Tolman, Organometallics, 13, 2855 (1994).
[15] M. Bovens, A. Togni, and L. M. Venanzi, J. Organomet. Chem., 451, C28 (1993).
[16] D. L. Christenson, C. J. Tokar, and W. B. Tolman, Organometallics, 14, 2148 (1995).
[17] M. H. Chisholm, N. W. Eilerts, J. C. Huffman, S. S. Iyer, M. Pacold, and K. Phomphrai, J. Am. Chem. Soc., 122, 11854 (2000).
[18] C. Kashima, H. Harada, I. Kita, I. Fukuchi, and A. Hosomi, Synthesis, 61 (1994).
[19] C. Kashima, I. Fukuchi, K. Takahashi, and A. Hosomi, Heterocycles, 38, 1407 (1994).
[20] D. Savoia, E. Tagliavini, C. Trombini, and A. UmmaniRonchi, J. Org. Chem., 46, 5344 (1981).

